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Key indicators

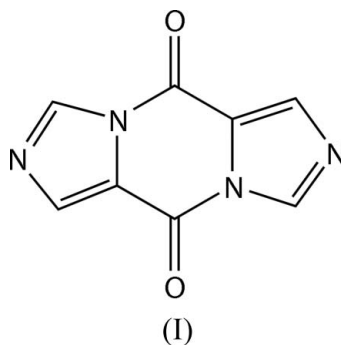
Single-crystal X-ray study
 $T = 173$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.041
 wR factor = 0.104
Data-to-parameter ratio = 6.5For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Diimidazo[1,5-*a*;1',5'-*d*]pyrazine-5,10-dione

The title compound, $\text{C}_8\text{H}_4\text{N}_4\text{O}_2$, was prepared to explore the synthesis of some 4-substituted imidazoles. The molecule is centrosymmetric and almost planar. In the crystal structure, centrosymmetric dimers are formed *via* $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds. The dimers are linked by further $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds, forming a three-dimensional network.

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Comment

As noted by Kasina & Nematollahi (1975), alkyl 4-imidazolecarboxylates and their derivatives are tedious to synthesize. They reported a simple synthesis based on the reaction of diimidazo[1,5-*a*;1',5'-*d*]pyrazine-5,10-dione, (I), with a number of reagents such as methanol, hydrazine, methylhydrazine and 1,1-dimethylhydrazine. Compound (I) (Fig. 1) was synthesized by the reaction of 4,5-imidazole-dicarboxylic acid with acetic anhydride.



The molecule is centrosymmetric and almost planar. The maximum deviation from the least-squares plane defined by the entire molecule is 0.010 (2) Å for atom N1. A search of the Cambridge Structural Database (Version 1.8, last update November 2006; Allen, 2002) revealed only one other compound containing the diimidazo[1,5-*a*;1',5'-*d*]pyrazine-5,10-dione unit, namely 3,8-bis(*p*-methylphenylamino)-1,6-dimethyldiimidazo[1,5-*a*;1',5'-*d*]pyrazine-5,10-dione, (II) (Burak *et al.*, 1992). A comparison of the bond distances and angles indicates some differences. For example, bond $\text{C4}=\text{O1}$ is 0.028 Å shorter, while bonds $\text{C3}-\text{C4}$ and $\text{N1}-\text{C3}$ are 0.024 and 0.026 Å, respectively, longer in (I) than in (II). For the remainder of the molecule the differences are between 1 and 4 s.u.

In the crystal structure of (I), centrosymmetric dimers are formed *via* $\text{C2}-\text{H2}\cdots\text{N2}^{\text{ii}}$ hydrogen bonds. These dimers are further linked by $\text{C1}-\text{H1}\cdots\text{N2}^{\text{i}}$ hydrogen bonds, forming a three-dimensional network. Details of the hydrogen bonding and symmetry codes are given in Table 1 and Fig. 2.

Experimental

Compound (I) was synthesized according to the procedure described by Kasina & Nematollahi (1975). Imidazole-4,5-dicarboxylic acid (2.0 g, 12.82 mmol) was introduced, under an atmosphere of nitrogen, into a 50 ml three-necked flask equipped with a magnetic stirring bar. An excess of acetic anhydride (20 ml) was added and the mixture stirred and heated under reflux at 433 K for 3 d. The dark solution obtained was evaporated under vacuum, giving a sticky solid which was washed with CCl_4 . When all the acetic acid had been eliminated, the residue was sublimed on heating at 463 K, using an oil bath and under vacuum. A pure yellow solid was obtained on the refrigerated finger. On recrystallization from CHCl_3 , pale-yellow rod-like X-ray-quality crystals of (I) were obtained (yield 0.42 g, 35%).

Crystal data

$\text{C}_8\text{H}_4\text{N}_4\text{O}_2$	$Z = 2$
$M_r = 188.15$	$D_x = 1.722 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 5.4047 (19) \text{ \AA}$	$\mu = 0.13 \text{ mm}^{-1}$
$b = 6.9777 (18) \text{ \AA}$	$T = 173 (2) \text{ K}$
$c = 9.678 (4) \text{ \AA}$	Rod, pale yellow
$\beta = 96.23 (3)^\circ$	$0.23 \times 0.13 \times 0.10 \text{ mm}$
$V = 362.8 (2) \text{ \AA}^3$	

Data collection

Stoe IPDS-2 diffractometer	471 independent reflections
ω and φ scans	338 reflections with $I > 2\sigma(I)$
Absorption correction: none	$R_{\text{int}} = 0.051$
1582 measured reflections	$\theta_{\text{max}} = 25.6^\circ$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0657P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.041$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.104$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 0.97$	$\Delta\rho_{\text{max}} = 0.19 \text{ e \AA}^{-3}$
471 reflections	$\Delta\rho_{\text{min}} = -0.16 \text{ e \AA}^{-3}$
73 parameters	Extinction correction: <i>SHELXL97</i>
All H-atom parameters refined	Extinction coefficient: 0.065 (16)

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C1}-\text{H1}\cdots\text{N2}^i$	0.97 (3)	2.49 (3)	3.440 (4)	166 (3)
$\text{C2}-\text{H2}\cdots\text{N2}^{ii}$	0.96 (3)	2.57 (3)	3.362 (4)	140 (3)

Symmetry codes: (i) $-x + 1, y - \frac{1}{2}, -z + \frac{3}{2}$; (ii) $-x + 1, -y + 1, -z + 1$.

Crystals of compound (I) were non-merohedral twins (matrix 1.00406, 0.00693, 0.00563, $-0.00918 - 1.00824, -0.00523, -0.38756, 0.01207, -1.00098$). The TWIN integration routine (*X-AREA*; Stoe & Cie, 2005) was used to obtain the combined data file. Approximately 32% of the reflections were overlapped and were eliminated from the final data used here. The H atoms were located in difference Fourier maps and refined isotropically; the C–H bond lengths are 0.96 (3) and 0.97 (3) \AA .

Data collection: *X-AREA* (Stoe & Cie, 2005); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2005); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

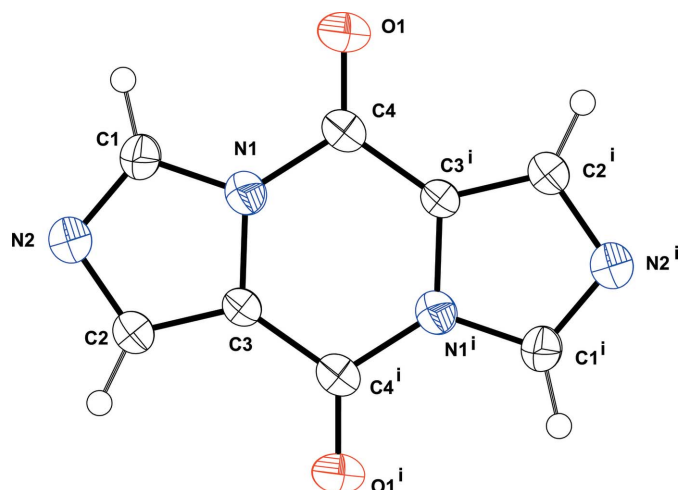


Figure 1

The molecular structure of (I), showing displacement ellipsoids drawn at the 50% probability level (arbitrary spheres for the H atoms). Atoms labelled with the suffix *i* are generated by the symmetry operation $(2 - x, -y, 1 - z)$.

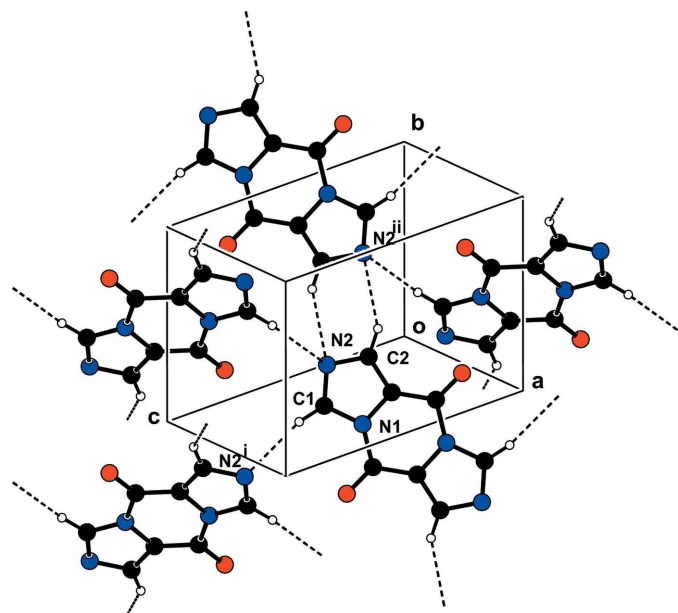


Figure 2

The crystal packing of (I), viewed perpendicular to (011). The C–H \cdots N hydrogen bonds are shown as dashed lines; symmetry codes correspond to those in Table 2.

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